

Examiner: David SAMPLE  
Art Unit: 1755  
Facsimile: 703-305-3599

Docket No.: NHL-SCT-21 US  
Serial No.: 09/758,903  
Telephone: 703-308-3825

In the Claims:

Please cancel claims 1-20, without prejudice.

Please add the following newly-presented claims:

*2C*

-21. A flat panel liquid-crystal display, such as for a laptop computer, the flat panel liquid-crystal display comprising one of: a twisted nematic display, a supertwisted nematic display, an active matrix liquid-crystal display, a thin film transistor display, and a plasma addressed liquid-crystal display, said flat panel liquid-crystal display comprising:  
backlight apparatus;  
a linear polarizer adjacent said apparatus configured to be a backlight;  
a first positive uniaxial retardation film adjacent said polarizer;  
a first negative retardation film adjacent said first positive uniaxial retardation film;  
a first orientation film adjacent said first negative retardation film;  
a liquid-crystal layer adjacent said first orientation film;  
a second orientation film adjacent said liquid-crystal layer;  
a second negative retardation film adjacent said second

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orientation film;

a second positive uniaxial retardation film adjacent said second retardation film;

a second linear polarizer adjacent said second positive uniaxial retardation film;

a first glass substrate being disposed between said first orientation film and said first negative retardation film;

a second glass substrate being disposed between said second orientation film and said second negative retardation film;

a first electrode being disposed between said first glass substrate and said first orientation film; and

a second electrode being disposed between said second glass substrates and said second orientation film;

said first and said second glass substrates comprising:

an alkali-free aluminoborosilicate glass;

said glass having a coefficient of thermal expansion  $\alpha_{20/300}$  of between  $2.8 \times 10^{-6}/K$  and  $3.8 \times 10^{-6}/K$ ;

said glass having the composition (in % by weight, based on oxide):

SiO <sub>2</sub>	> 58 - 65
B <sub>2</sub> O <sub>3</sub>	> 6 - 11.5
Al <sub>2</sub> O <sub>3</sub>	> 21 - 25
MgO	4 - 8

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CaO 0 - 8  
SrO 2.6 - < 8  
BaO 0 - < 0.5  
ZnO 0 - 2;

said glass being configured to be resistant to thermal  
shock;

said glass being configured to having a high transparency  
over a broad spectral range in the visible and ultra violet  
ranges; and

said glass being configured to be free of bubbles, knots,  
inclusions, streaks, and surface undulations.--

--22. The flat panel liquid-crystal display according to  
claim 21, wherein:

said glass comprises at least one of (a.), (b.), (c.), (d.),  
(e.), and (f.), where (a.), (b.), (c.), (d.), (e.), and (f.) are:

(a.) more than 8% by weight of  $B_2O_3$ ;

(b.) one of: more than 18% by weight of  $Al_2O_3$ , at least  
20.5% by weight of  $Al_2O_3$ , and at least 21% by weight of  $Al_2O_3$ ;

(c.) additionally (in % by weight):

$ZrO_2$  0 - 2  
 $TiO_2$  0 - 2  
with  $ZrO_2 + TiO_2$  0 - 2

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As2O3 0 - 1.5  
Sb2O3 0 - 1.5  
SnO2 0 - 1.5  
CeO2 0 - 1.5  
Cl^- 0 - 1.5  
F^- 0 - 1.5  
SO4^2- 0 - 1.5

with As2O3 + Sb2O3 + SnO2 + CeO2

+ Cl^- + F^- + SO4^2- 0 - 1.5;

*B*  
(d.) a glass in which arsenic oxide, antimony oxide, and inherent impurities are minimized;

(e.) a float glass; and

(f.) one of (i.), (ii.), and (iii.):

(i.) a coefficient of thermal expansion  $\alpha_{20/300}$  of between  $2.8 \times 10^{-6}/K$  to  $3.6 \times 10^{-6}/K$ ;

(ii.) a glass transition temperature  $T_g$  of  $> 700^\circ C$ ; and

(iii.) a density  $\rho$  of  $< 2.600 \text{ g/cm}^3$ .--

--23. The flat panel liquid-crystal display according to claim 21, wherein:

said glass comprises (a.), (b.), (c.), (d.), (e.), and (f.), where (a.), (b.), (c.), (d.), (e.), and (f.) are:

(a.) more than 8% by weight of B2O3;

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(b.) one of: more than 18% by weight of  $\text{Al}_2\text{O}_3$ , at least 20.5% by weight of  $\text{Al}_2\text{O}_3$ , and at least 21% by weight of  $\text{Al}_2\text{O}_3$ ;

(c.) additionally (in % by weight):

$\text{ZrO}_2$	0 - 2
$\text{TiO}_2$	0 - 2
with $\text{ZrO}_2 + \text{TiO}_2$	0 - 2
$\text{As}_2\text{O}_3$	0 - 1.5
$\text{Sb}_2\text{O}_3$	0 - 1.5
$\text{SnO}_2$	0 - 1.5
$\text{CeO}_2$	0 - 1.5
$\text{Cl}^-$	0 - 1.5
$\text{F}^-$	0 - 1.5
$\text{SO}_4^{2-}$	0 - 1.5

with  $\text{As}_2\text{O}_3 + \text{Sb}_2\text{O}_3 + \text{SnO}_2 + \text{CeO}_2$   
 $+ \text{Cl}^- + \text{F}^- + \text{SO}_4^{2-}$  0 - 1.5;

(d.) a glass in which arsenic oxide, antimony oxide, and inherent impurities are minimized;

(e.) a float glass; and

(f.) one of (i.), (ii.), and (iii.):

(i.) a coefficient of thermal expansion  $\alpha_{20/300}$  of between  $2.8 \times 10^{-6}/\text{K}$  to  $3.6 \times 10^{-6}/\text{K}$ ;

(ii.) a glass transition temperature  $T_g$  of  $> 700^\circ\text{C}$ ; and

(iii.) a density  $\rho$  of  $< 2.600 \text{ g/cm}^3$ ...

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--24. A glass substrate for a flat panel liquid-crystal display, such as for a laptop computer, the flat panel liquid-crystal display including a twisted nematic display, a supertwisted nematic display, an active matrix liquid-crystal display, a thin film transistor display, and a plasma addressed liquid-crystal display, said substrate comprising:

an alkali-free aluminoborosilicate glass;  
said glass having a coefficient of thermal expansion  $\alpha_{20/300}$  of between  $2.8 \times 10^{-6}/K$  and  $3.8 \times 10^{-6}/K$ ;  
said glass having the composition (in % by weight, based on oxide):

SiO <sub>2</sub>	> 58 - 65
B <sub>2</sub> O <sub>3</sub>	> 6 - 11.5
Al <sub>2</sub> O <sub>3</sub>	> 14 - 25
MgO	4 - 8
CaO	0 - < 2
SrO	> 0.5 - < 4
BaO	0 - < 0.5
ZnO	0 - 2;

    said glass being configured to be resistant to thermal shock;

    said glass being configured to having a high transparency over a broad spectral range in the visible and ultra violet

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ranges; and

      said glass being configured to be free of bubbles, knots,  
 inclusions, streaks, and surface undulations. --

-- 25.    The glass substrate according to claim 24,  
 wherein:

      said glass comprises at least one of (a.), (b.), (c.), (d.),  
 (e.), and (f.), where (a.), (b.), (c.), (d.), (e.), and (f.) are:

      (a.) more than 8% by weight of  $B_2O_3$ ;

      (b.) one of: more than 18% by weight of  $Al_2O_3$ , at least  
 20.5% by weight of  $Al_2O_3$ , and at least 21% by weight of  $Al_2O_3$ ;

      (c.) additionally (in % by weight):

$ZrO_2$    0 - 2

$TiO_2$    0 - 2

      with  $ZrO_2$  +  $TiO_2$    0 - 2

$As_2O_3$    0 - 1.5

$Sb_2O_3$    0 - 1.5

$SnO_2$    0 - 1.5

$CeO_2$    0 - 1.5

$Cl^-$    0 - 1.5

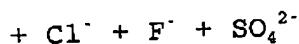
$F^-$    0 - 1.5

$SO_4^{2-}$    0 - 1.5

      with  $As_2O_3$  +  $Sb_2O_3$  +  $SnO_2$  +  $CeO_2$

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0 - 1.5;

(d.) a glass in which arsenic oxide, antimony oxide, and inherent impurities are minimized;

(e.) a float glass; and

(f.) one of (i.), (ii.), and (iii.):

(i.) a coefficient of thermal expansion  $\alpha_{20/300}$  of between  $2.8 \times 10^{-6}/\text{K}$  to  $3.6 \times 10^{-6}/\text{K}$ ;

(ii.) a glass transition temperature  $T_g$  of  $> 700^\circ\text{C}$ ; and

(iii.) a density  $\rho$  of  $< 2.600 \text{ g/cm}^3$ .--

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 B --26. The glass substrate according to claim 24,

wherein:

said glass comprises (a.), (b.), (c.), (d.), (e.), and (f.), where (a.), (b.), (c.), (d.), (e.), and (f.) are:

(a.) more than 8% by weight of  $\text{B}_2\text{O}_3$ ;

(b.) one of: more than 18% by weight of  $\text{Al}_2\text{O}_3$ , at least 20.5% by weight of  $\text{Al}_2\text{O}_3$ , and at least 21% by weight of  $\text{Al}_2\text{O}_3$ ;

(c.) additionally (in % by weight):

$\text{ZrO}_2$

0 - 2

$\text{TiO}_2$

0 - 2

with  $\text{ZrO}_2 + \text{TiO}_2$

0 - 2

$\text{As}_2\text{O}_3$

0 - 1.5

$\text{Sb}_2\text{O}_3$

0 - 1.5

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$\text{SnO}_2$  0 - 1.5  
 $\text{CeO}_2$  0 - 1.5  
 $\text{Cl}^-$  0 - 1.5  
 $\text{F}^-$  0 - 1.5  
 $\text{SO}_4^{2-}$  0 - 1.5

with  $\text{As}_2\text{O}_3$  +  $\text{Sb}_2\text{O}_3$  +  $\text{SnO}_2$  +  $\text{CeO}_2$   
 +  $\text{Cl}^-$  +  $\text{F}^-$  +  $\text{SO}_4^{2-}$  0 - 1.5;

*B*  
 (d.) a glass in which arsenic oxide, antimony oxide, and inherent impurities are minimized;

(e.) a float glass; and

(f.) one of (i.), (ii.), and (iii.):

(i.) a coefficient of thermal expansion  $\alpha_{20/300}$  of between  $2.8 \times 10^{-6}/\text{K}$  to  $3.6 \times 10^{-6}/\text{K}$ ;

(ii.) a glass transition temperature  $T_g$  of  $> 700^\circ\text{C}$ ; and

(iii.) a density  $\rho$  of  $< 2.600 \text{ g/cm}^3$ .--

--27. A glass comprising:

a substantially alkali-free aluminoborosilicate glass;

said glass having a coefficient of thermal expansion  $\alpha_{20/300}$  of between  $2.8 \times 10^{-6}/\text{K}$  and  $3.8 \times 10^{-6}/\text{K}$ ;

said glass having the composition (in % by weight, based on oxide):

$\text{SiO}_2$  > 58 - 65

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$B_2O_3$	> 6 - 11.5
$Al_2O_3$	> 14 - 25
MgO	4 - 8
CaO	0 - 8
SrO	2.6 - < 4
BaO	0 - < 0.5
with SrO + BaO	> 3
ZnO	0 - 2. --

*B*  
 --28. The glass according to claim 27, wherein:  
 said glass is configured to be resistant to thermal shock;  
 said glass is configured to having a high transparency over  
 a broad spectral range in the visible and ultra violet ranges;  
 and

said glass is configured to be free of bubbles, knots,  
 inclusions, streaks, and surface undulations...

--29. The glass according to claim 28, wherein:  
 said glass comprises more than 8% by weight of  $B_2O_3$ ...

--30. The glass according to claim 29, wherein:  
 said glass comprises one of (i.) and (ii.):  
 (i.) more than 18% by weight of  $Al_2O_3$ ; and

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(ii.) at least 20.5% by weight of  $\text{Al}_2\text{O}_3$ . --

--31. The glass according to claim 30, wherein said glass comprises at least 21.5% by weight of  $\text{Al}_2\text{O}_3$ . --

--32. The glass according to claim 31, wherein: said glass additionally comprises (in % by weight):

$\text{ZrO}_2$	0 - 2
$\text{TiO}_2$	0 - 2
with $\text{ZrO}_2 + \text{TiO}_2$	0 - 2
$\text{As}_2\text{O}_3$	0 - 1.5
$\text{Sb}_2\text{O}_3$	0 - 1.5
$\text{SnO}_2$	0 - 1.5
$\text{CeO}_2$	0 - 1.5
$\text{Cl}^-$	0 - 1.5
$\text{F}^-$	0 - 1.5
$\text{SO}_4^{2-}$	0 - 1.5; and
with $\text{As}_2\text{O}_3 + \text{Sb}_2\text{O}_3 + \text{SnO}_2 + \text{CeO}_2$	
$+ \text{Cl}^- + \text{F}^- + \text{SO}_4^{2-}$	0 - 1.5. --

--33. The glass according to claim 32, wherein: said glass comprises a glass in which arsenic oxide, antimony oxide, and inherent impurities are minimized. --

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--34. The glass according to claim 33, wherein:  
 said glass comprises a float glass.--

--35. The glass according to claim 34, wherein:  
 said glass has one of (i.), (ii.), and (iii.):  
 (i.) a coefficient of thermal expansion  $\alpha_{20/300}$  of between  $2.8 \times 10^{-6}/K$  to  $3.6 \times 10^{-6}/K$ ;  
 (ii.) a glass transition temperature  $T_g$  of  $> 700^{\circ}C$ ; and  
 (iii.) a density  $\rho$  of  $< 2.600 \text{ g/cm}^3$ .--

--36. The glass according to claim 27, wherein:  
 said glass comprises at least one of (a.), (b.), (c.), (d.), (e.), and (f.), where (a.), (b.), (c.), (d.), (e.), and (f.) are:  
 (a.) more than 8% by weight of  $B_2O_3$ ;  
 (b.) one of: more than 18% by weight of  $Al_2O_3$ , at least 20.5% by weight of  $Al_2O_3$ , and at least 21% by weight of  $Al_2O_3$ ;  
 (c.) additionally (in % by weight):

$ZrO_2$	0 - 2
$TiO_2$	0 - 2
with $ZrO_2 + TiO_2$	0 - 2
$As_2O_3$	0 - 1.5
$Sb_2O_3$	0 - 1.5
$SnO_2$	0 - 1.5

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$\text{CeO}_2$

0 - 1.5

$\text{Cl}^-$

0 - 1.5

$\text{F}^-$

0 - 1.5

$\text{SO}_4^{2-}$

0 - 1.5

with  $\text{As}_2\text{O}_3$  +  $\text{Sb}_2\text{O}_3$  +  $\text{SnO}_2$  +  $\text{CeO}_2$   
 +  $\text{Cl}^-$  +  $\text{F}^-$  +  $\text{SO}_4^{2-}$

0 - 1.5;

(d.) a glass in which arsenic oxide, antimony oxide, and inherent impurities are minimized;

(e.) a float glass; and

(f.) one of (i.), (ii.), and (iii.):

(i.) a coefficient of thermal expansion  $\alpha_{20/300}$  of between  $2.8 \times 10^{-6}/\text{K}$  to  $3.6 \times 10^{-6}/\text{K}$ ;

(ii.) a glass transition temperature  $T_g$  of  $> 700^\circ\text{C}$ ; and

(iii.) a density  $\rho$  of  $< 2.600 \text{ g/cm}^3$ ...

--37. The glass according to claim 27, wherein:

said glass is configured as a glass substrate in combination in or with a flat panel liquid-crystal display, such as for a laptop computer, the flat panel liquid-crystal display including a twisted nematic display, a supertwisted nematic display, an active matrix liquid-crystal display, a thin film transistor display, and a plasma addressed liquid-crystal display.--

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--38. The glass according to claim 37, wherein:  
said flat panel liquid-crystal display comprises:  
backlight apparatus;  
a linear polarizer adjacent said apparatus configured to be  
a backlight;  
a first positive uniaxial retardation film adjacent said  
polarizer;  
a first negative retardation film adjacent said film;  
a first orientation film adjacent said retardation film;  
a liquid-crystal layer adjacent said first orientation film;  
a second orientation film adjacent said liquid-crystal  
layer;  
a second negative retardation film adjacent said second  
orientation film;  
a second positive uniaxial retardation film adjacent said  
second retardation film;  
a second linear polarizer adjacent said second retardation  
film;  
a first glass substrate being disposed between said first  
orientation film and said first negative retardation film;  
a second glass substrate being disposed between said second  
orientation film and said second negative retardation film;  
a first electrode being disposed between said first glass

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substrate and said first orientation film; and  
a second electrode being disposed between said second glass  
substrates and said second orientation film. --

--39. The glass according to claim 27, wherein:  
said glass is configured as a glass substrate in combination  
in or with a thin-film photovoltaic device, including a thin-film  
solar cell. --

--40. The glass according to claim 39, wherein,  
said thin-film photovoltaic device comprises:  
said glass substrate;  
a transparent conductive oxide film disposed on said  
substrate;  
an insulating buffer layer disposed atop said transparent  
conductive oxide film;  
said film being disposed between said glass substrate and  
said buffer layer and being configured to be a front contact  
current collector;  
a first semiconductor layer disposed upon said buffer layer;  
a second semiconductor layer disposed upon said first  
semiconductor layer to form a heterojunction;  
a first electrical contact disposed upon said second